NEW CLERODANES FROM SYMPHIOPAPPUS ITATIAYENSIS*

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Abstract—The isolation and structure determinations of two new ent-cis-clerodane dilactones from Symphiopappus itatiayensis are reported.

In the course of our work on constituents of Brazilian Compositae, we have isolated from the hexane-EtOAc (10:1) extract of Symphiopappus itatiayensis, tribe Eupatorieae, Disynaphia group [2], two new clerodane dilactones whose structures 1a and 2 are reported herewith.

The four oxygen atoms of 1a, mp 150°, C₂₀H₂₆O₄ (elementary analysis and high resolution MS) were those of two conjugated lactone functions as indicated by the empirical formula, the absence of hydroxyl groups and by IR bands at 1750 (very strong) and 1650 cm⁻¹. NMR analysis showed that these were included in partial structures A and B (numbering as in final formula). Chemical shifts of C-13, C-14, C-15, and C-16 at 170.1, 115.2, 172.3, and 72.9 ppm, with H-14 at 5.85 being longrange coupled (J = 1.5) to the two proton doublet of H-16 at 4.75 ppm, were essentially identical with those found in an ent-trans-clerodane from Baccharis trimera [3] containing a similar lactone system (also see [4]); likewise, chemical shifts and coupling constants of H-2 (buried in a four proton multiplet at 2.35 ppm), H-3 (t at 6.83 ppm, J = 3.5 Hz) and H-6 (dd at 4.32 ppm, J = 10.5, 7 Hz) were essentially identical with H-2, H-3, and H-6 of 1b from Solidago arguta [5]. Signals of C-3, C-4, and C-19, C-6 were located at 133.3, 135.9, 169.5, and 84.8 ppm. respectively. Spin-decoupling experiments permitted identification of H-7a and H-7b as two multiplets at 1.92 and 1.32 ppm, each being coupled to a multiplet at 1.52 ppm (H-8) which was in turn responsible for splitting the signal of H-17 at 0.85 ppm.

22.0, 24.5, 30.5, 34.7 and 35.0 ppm, two upfield methine doublets at 32.0 (C-8) and 41.4 (C-10) and two quaternary carbons at 30.0 and 30.4 (C-5 and C-9), as well as three quartets at 15.2, 16.2, and 17.6 ppm completed the argument leading to gross structure 1a, which was supported by the appearance in the high resolution MS of important peaks at m/e 219 (C₁₄H₁₉O₂) corresponding to loss of the six carbon side chain and 111 (C₆H₇O₂) corresponding to the side chain itself.

Because of the coincidence of the relevant NMR signals with those exhibited by 1b whose structure has been established by X-ray crystallography [6], we deduced that the new dilactone was the ent-cis-clerodane 1a. Several attempts to correlate 1a and 1b by reduction of 1a with (iso-Bu), AlH to a diol previously [5] obtained from 1b resulted in mixtures which could not be separated satisfactorily; however, a solvent shift study of the two substances (Table 2) showed that $\Delta \delta$ s for H-3, H-6, and H-18 were essentially identical. Since the somewhat larger differences in $\Delta\delta$ for H-17 and H-20 are easily attributable to the proximity of these protons to the lactone and furanoid side chains of 1a and 1b respectively, we conclude that the relative stereochemistry of the two substances is indeed the same. The absolute stereochemistry as depicted is consonant with the strongly negative Cotton effect $([\theta]_{250} - 23900).$

A second substance 2, mp 229–230°, $C_{20}H_{28}O_5$ (high resolution MS) which was isolated in very small amounts only differed from 1a in possessing a saturated γ -lactone

indicates quaternary carbon

Two methyl singlets at 1.26 and 0.77 ppm and consideration of the ¹³C-NMR spectrum which exhibited upfield methylene triplets (including those of C-2 and C-7) at

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function in the side chain (MS and NMR, see Table 1) and an extra hydroxyl group (IR) which was located on C-2 (m at 4.63 ppm which was spin coupled to H-3 and to signals at 2.26 and 1.91 ppm—H-1 a, b). The similarity of chemical shifts and coupling constants where appropriate (Table 1) shows that the stereochemistry of 2 corresponds to that of 1a as does its absolute configuration ($[\theta]_{250}$ -6500). The magnitude of the coupling constants involving H-2 requires that the hydroxyl group on C-2 be

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Table 1. 270 MHz NMR spectra of compounds 1a and 2*

	1a	2		
H-1	+	2.26 m		
		1.91 m		
H-2	$2.35m_{+}^{*}$	$4.63 ddd (J_{1a, 2} = 9, J_{1b, 2} = 6)$		
H-3	$6.83\ t\ (3.5)$	6.79 d (3)		
H-6	4.32 dd (10.5, 7)	4.25 (10.5, 7.5)		
H-7a	1.97 m	1.91 m		
7b	1.32 m	1.34m		
H-8	1.52 m	1,34		
H-10	†	1.91 m		
H-11	†	+		
H-12	†	†		
H-13		2.44 m		
H-14	5.84 m	2.64 dd (16.6, 8)		
		2.16 (16.6, 8)		
H-16	4.75 d (1.5)‡	4.40 dd (9, 6)		
	. , ,	3.90 dd (9, 6)		
H-17§	0.85 d (6.5)	0.80 d (6.5)		
H-18§		1,27		
H-20§	0.77	0.69		

^{*} Run in CDCl₃. Chemical shifts in ppm. Coupling constants in hertz. Unmarked signals are singlets.

quasiequatorial. Its configuration depends on the conformation of ring A; if the latter is a half-chair, the hydroxyl group must be β , if a half-boat, α . While inspection of the models does not permit a facile choice between the two possibilities, the crystal structure of 1b [6] indicates a half-chair conformation for ring A of 1b in the solid state and presumably therefore also for ring A of 2. Consequently, the hydroxyl group of 2 should be β , subject to the proviso that the conformations in solution and in the solid state are the same.

EXPERIMENTAL

Aerial parts of Symphiopappus itatiayensis (Hieron.) King & Robinson (5.2 kg) collected by Dr. Hermogenes de Freitas Leitão Filho in Campos do Jordão (Pico de Itapeva), São Paulo State, Brazil in March 1974 were extracted with hexane–EtOAc (10:1), giving 129 g of extract which was worked up in the usual manner. The crude gum (30 g) was chromatographed over 270 g Si gel. 250 ml fractions were eluted in the following order: 1–7 (hexane), 8–12 (hexane–EtOAc, 20:1), 13–15 (hexane–EtOAc, 10:1), 16–17 (5:1), 18–50 (3.3:1), 51–55 (2:1), 56–60 (1.4:1), 61–64 (EtOAc), 65–66 (EtOAc–EtOH, 21:1), 67–78 (EtOAc–EtOH, 10:1), 79–84 (3.3:1), 85–87 (2:1), 83–93 (EtOH). Fractions 41 and 42 contained 0.3 g of crude 1a which was purified by

Table 2. Solvent shifts of compounds 1a and 1b*

	CDCl ₃		C_6D_6		$\Delta\delta$	
	1a	1b	1a °	1b	1a	1 b
H-3	6.83	6.85	6.67	6.65	+ 0.20	+0.20
H-6	4.32	4.35	3.88	3.87	+0.44	+0.48
H-17	0.85	0.88	0.32	0.49	+0.53	+0.39
H-18	1.26	1.27	0.93	0.94	+0.33	+0.34
H-20	0.77	0.74+	0.25	0.34	+0.52	+0.40

^{*} Run at 270 MHz. Values in ppm.

In all other respects the spectrum was the same as that reported by McCrindle et al. [5].

TLC (Si gel PF 254–366; EtOAc– C_6H_6 , 1:3), mp 150°, IR $v_{\rm max}^{\rm KBr}$ cm⁻¹: 1780 (weak), 1750 (very strong), 1695 (weak) and 1640, $\left[\alpha\right]_{\rm D}$ 9.62° (c 0.97, CHCl $_3$), CD curve $\left[\theta\right]_{2.50}^{\rm CHCl}$ – 23900, UV $\lambda_{\rm max}^{\rm MeOH}$ 216 nm (ϵ 10300), significant peaks in the high resolution MS at m/e (rel. int.) 330 (M⁺), 315 (M⁺ – CH $_3$, 16.9), 297 (M⁺ – CH $_3$ – H $_2$ O, 13.5), 219 (C $_{14}$ H $_{19}$ O $_2$, 25.3), 111 (C $_6$ H $_7$ O $_2$, 24.9). (Calc. for C_{20} H $_{26}$ O $_4$: C, 72.70; H, 7.93; O, 19.37; MW, 330.1830. Found: C, 72.27; H, 7.89; O, 20.02%; MW (MS), 330.1845).

Fraction 58 contains 12 mg of solid which gave after purification by TLC (hexane–EtOAc, 3:1) and elution of the major spot with CHCl₃–MeOH (50:1) 7 mg of 2, mp 229–230 , IR $v_{\rm max}^{\rm KBr}$ cm⁻¹: 3520, 1760 (very strong, sh at 1720, and 1630, CD curve $[\theta]_{256}$ – 6500, UV $z_{\rm max}^{\rm MeOH}$ 213 nm (ε 7700) significant peaks in the high resolutions MS at me (rel. int.) 348 (M⁺, 0.5) 333 (M⁺ – CH₃, 100), 315 (M⁺ – CH₃ – H₂O, 77.5). (Calc for $C_{20}H_{28}O_5$: MW, 348.1935: Found: MW (MS), 348.1956).

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[†] Obscured signal.

[‡] Intensity two protons.

[§] Intensity three protons.

^{*}This value differed significantly from the value of 0.59 ppm given in ref. [5].